TRANSMITTAL LETTER TO THE UNITED STATES

U.S. DEPARTMENT OF COMMERCE PATENT AND TRADEMARK OFFICE

Customized FORM PTO-1390

ATTORNEY DOCKET NO. P07223US00/LRP

	ECTED OFFICE (DO/EO/US) FILING UNDER 35 U.S.C. 371	U.S. APPLICATION NO. 0 911/1085351358									
INTERNATIONAL APPLICATION N	O. INTERNATIONAL FILING DATE 22 November 1999	PRIORITY DATE CLAIMED 24 November 1998									
PCT/JP99/06502	SOLUTION, ETCHED ARTICLE AND ME										
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Applicant herewith submits to the OS D	esignated/Elected Office (DO/EO/O3) the follow	ing terns and other information.									
MICROSOLO MAMA	n of items concerning a filing under 35 U.S.										
2. This is a SECOND or SU	SEQUENT submission of items concerning	g a filing under 35 USC 371.									
	in national examination procedures (35 USC ation of the applicable time limit set in 35 Usc										
X 4. A proper Demand for Inter claimed priority date.											
5. A copy of the International	Application as filed (35 U.S.C. 371 (c)(2))										
a. is transmitted herewit	h (required only if not transmitted by the Int	ernational Bureau).									
X b. has been transmitted	by the International Bureau.										
c. is not required, as the	application was filed in the United States Re	eceiving Office (RO/US).									
X 6. A translation of the Intern	ational Application into English (35 U.S.C.	371(c)(2)).									
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	ith (required only if not transmitted by the In	nternational Bureau).									
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d. have not been made a											
8. A translation of the amend	ments to the claims under PCT Article 19 (3:	5 U.S.C. 371(c)(3)).									
X 9. An oath or declaration of t	he inventor(s) (35 U.S.C. 371(c)(4)).										
10. A translation of the annexe	s to the Int'l Prelim. Exam. Report under PC	T Article 36 (35 U.S.C. 371(c)(5)).									
Items 11 to 20 below concern	document(s) or information included:										
	e Statement under 37 C.F.R. 1.97 and 1.98.										
	for recording. A separate cover sheet in compliant	ance with 37 CFR 3.28 and 3.31 is included.									
X 13. A First preliminary amer											
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Date: May 22, 2001

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Independent Claims	1 - 03 =	0		X \$80 =	=	\$ 0					
Multiple Depend	ent Claim(s) (if applie	cable)		+ \$270 =	=	\$ 0					
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Applicant claims small entity status. See 37 CFR 1.27. The fees indicated above are reduced by ½.											
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Processing fee of \$130 for furnishing the English translation later than from the earliest claimed priority date (37 CFR 1.492(f)). SUBTOTAL = \$860 [] 20 mos.											
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At the address (below)	of CUSTOMER NO	1 4444221									
LARSON &	& TAYLOR, PLC		NAME: Douglas E. Jackson REG. NO.: 28,518								
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IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

Patent

In re patent application of: KEZUKA ET AL.

Serial No.: Unassigned

Examiner: Unassigned

Filed: May 22, 2001

Art Unit: Unassigned

For: ETCHING SOLUTION, ETCHED ARTICLE AND

Docket No.: P07223US00/LRP

METHOD FOR ETCHED ARTICLE

PRELIMINARY AMENDMENT

Assistant Commissioner of Patents

Washington, D.C. 20231

SIR:

Prior to examination, please amend the above-identified application as follows:

IN THE CLAIMS

A clean version of all amended claim 14 is provided herewith in **Attachment A**. It will be noted that claim 14 has been amended relative to the previously provided version submitted with the application shown by the marked up version thereof in **Attachment B** provided herewith.

REMARKS

The present Amendment is made to eliminate multiple dependency in the claims.

Respectfully submitted,

Date: May 22, 2001

By: Douglas E. Jackson Jr. Registration No.: 28,518

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ATTACHMENT A

Clean Replacement/New Claims

Following herewith is a marked up copy of each rewritten claim.

15. (AMENDED) A method for producing an etched article by etching an article to be etched with the etching solution as defined in claim 1.

ATTACHMENT B

Marked Up Replacement Claims

Following herewith is a marked up copy of each rewritten claim.

15. (AMENDED) A method for producing an etched article by etching an article to be etched with the etching solution as defined in any of claims 1-14 claim 1.

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DESCRIPTION

ETCHING SOLUTION, ETCHED ARTICLE

AND METHOD FOR ETCHED ARTICLE

TECHNICAL FIELD

The present invention relates to an etching solution, a method for producing an etched article and an etched article produced by the method, more specifically, an etching solution and a method for producing an etched article for selectively etching a doped oxide film, particularly BSG or BPSG relative to an undoped oxide film, particularly THOX, and an etched article produced by the method.

BACKGROUND ART

Conventionally, as etchants for silicon wafers and the like have been used buffered hydrofluoric acids comprising HF (50% by weight) and NH $_4$ F (40% by weight) at such a ratio that can achieve a desired etch rate.

However, the buffered hydrofluoric acids etch not only doped oxide films such as BSG films, BPSG films, phosphosilicate glass (PSG) films, arsenic silicate glass (AsSG) films and the like, but also undoped oxide films such as USG including TEOS (oxide obtained by CVD method using tetraethoxysilane gas) films, THOX and the like. Therefore, the buffered hydrofluoric acids can not selectively etch the doped oxide films.

An object of the present invention is to provide an etching solution and an etching method for selectively etching oxide films doped with impurities relative to TEOS and THOX.

5 <u>DISCLOSURE OF INVENTION</u>

The present invention relates to the items 1-16 listed below.

- Item 1: An etching solution comprising hydrofluoric acid, wherein an a ratio of etch rate of a boron silicate glass
- 10 film (BSG) or boron phosphosilicate glass / an etch rate of a thermal oxide film (THOX) at 25°C is 10 or higher.
 - Item 2: The etching solution according to item 1, wherein a solvent in the etching solution has a relative dielectric constant of 61 or lower.
- 15 Item 3: The etching solution according to item 1, the solution containing at least one member selected from the group consisting of an organic acid and an organic solvent having a hetero atom.
- Item 4: The etching solution according to item 1, the

 solution containing (i) water and (ii) at least one
 member selected from the group consisting of an organic
 acid and an organic solvent having a hetero atom, the
 water being contained in a concentration of 70% by weight
 or lower.
- 25 Item 5: The etching solution according to item 1,

wherein the weight ratio of HF: isopropyl alcohol: water is 0.1-50% by weight: 30-99% by weight: 0-70% by weight.

Item 6: The etching solution according to item 1,

wherein the weight ratio of HF: acetic acid: water is

0.1-50% by weight: 30-99.9% by weight: 0-70% by weight.

Item 7: The etching solution according to item 1,

wherein the weight ratio of HF: tetrahydrofuran: water

is 0.1-50% by weight: 30-99.9% by weight: 0-70% by

weight.

Item 8: The etching solution according to item 1, wherein the weight ratio of HF: acetone: water is 0.1-50% by weight: 30-99.9% by weight: 0-70% by weight.

Item 9: The etching solution according to item 1,

- wherein the weight ratio of HF: methanol: water is 0.1-50% by weight: 30-99.9% by weight: 0-70% by weight.

 Item 10: The etching solution according to item 1,

 wherein the weight ratio of HF: ethanol: water is 0.1-50% by weight: 30-99.9% by weight: 0-70% by weight.
- 20 Item 11: The etching solution according to item 1, the solution comprising an inorganic acid.

 Item: 12 The etching solution according to item 11, wherein the inorganic acid has a pKa value at 25°C of 2 or lower.
- 25 Item 13: The etching solution according to item 11,

wherein the weight ratio of HF: HCl: water is 0.01-50% by weight: 1-36% by weight: 0-99% by weight.

Item 14: The etching solution according to item 11, wherein the weight ratio of HF : HNO_3 : water is 0.01-50%

by weight: 1-70% by weight: 0-99% by weight.

Item 15: A method for producing an etched article by etching an article to be etched with the etching solution

as defined in any of items 1-14.

Item 16: An etched article which is obtainable by the 10 method of item 15.

According to the etching solution of the invention, the ratio of BSG etch rate / THOX etch rate and/or the ratio of BPSG etch rate / THOX etch rate at 25°C is/are 10 or higher, preferably 20 or higher, more preferably 50 or higher, particularly 100 or higher.

In case of using TEOS instead of THOX, the ratio of BSG etch rate / TEOS etch rate and/or the ratio of BPSG etch rate / TEOS etch rate at 25°C is/are 5 or higher, preferably 10 or higher, more preferably 50 or higher, particularly 100 or higher.

The etch rate of the etching solution of the invention can be calculated as the difference in thickness of a film (BSG; BPSG; THOX; TEOS and like USG, etc.) before and after etching divided by etch time.

The water content is not higher than 70% by

weight, preferably not higher than 30% by weight, more preferably about 30-5% by weight. The relative dielectric constant of the etching solution expresses an arithmetic mean of the relative dielectric constants of the components of the etching solutions other than the HF and inorganic acid.

Preferable examples of the inorganic acid include inorganic acids having a pKa value at 25°C of 2 or lower, for example, hydrochloric acid (pKa =-8), nitric acid (pKa=-1.8), hydrobromic acid (pKa=-9), hydroiodic acid (pKa=-10) and perchloric acid (a pKa-unmeasurably strong acid).

Examples of the organic acid include acetic acid (relative dielectric constant: 6.15 (20°C)), propionic acid (relative dielectric constant: 3.4 (40°C)), 15 butyric acid (relative dielectric constant: 2.97(20°C)), isobutyric acid (relative dielectric constant: 2.73(40°C)), valeric acid, caproic acid (relative dielectric constant: 2.63(71°C)), caprylic acid (relative 20 dielectric constant: 2.45(20°C)), monochloroacetic acid (relative dielectric constant: 21 (20°C)), dichloroacetic acid (relative dielectric constant: 8.08(20°C)), trichloroacetic acid (relative dielectric constant: 4.6 (60°C)), monofluoroacetic acid, difluoroacetic acid, 25 trifluoroacetic acid, α -chlorobutyric acid, β - chlorobutyric acid, \(\gamma \)-chlorobutyric acid, lactic acid (relative dielectric constant: 22(70°C)), glycolic acid, pyruvic acid, glyoxalic acid, acrylic acid and like monocarboxylic acids, methanesulfonic acid,

toluenesulfonic acid and like sulfonic acids, oxalic acid, succinic acid, adipic acid, tartaric acid, citric acid and like polycarboxylic acids.

Examples of the organic solvent having a hetero atom include methanol (relative dielectric constant: 32.6

- 10 (25°C)), ethanol (relative dielectric constant: 24.6 (25°C)), isopropanol (IPA, relative dielectric constant:
 - 19.9 (25°C)), 1-propanol (relative dielectric constant:
 - 22.2 (25°C)), 1-butanol (relative dielectric constant:
 - 17.1 (25°C)), 2-butanol (relative dielectric constant:
- 15 15.5 (19°C)), t-butanol (relative dielectric constant:
 - 11.4 $(19^{\circ}C)$), 2-methyl-1-propanol (relative dielectric

constant: 17.95 (20°C)), 1-pentanol (relative dielectric

constant: 13.9 (25°C)), 1-hexanol (relative dielectric

constant: 13.3 (25°C)), 1-heptanol, 4-heptanol, 1-octanol

- 20 (relative dielectric constant: 10.34 (20°C)), 1
 - nonylalcohol, 1-decanol, 1-dodecanol and like alcohols;
 - ethylene glycol (relative dielectric constant: 37.7
 - (20°C)), 1,2-propanediol (relative dielectric constant:
- 32.0 (20°C)), 2,3-butanediol, glycerin (relative
- 25 dielectric constant: 42.5 (25°C)) and like polyols,

acetone (relative dielectric constant: 20.7 (25°C)), acetylacetone, methyl ethyl ketone (relative dielectric constant: 18.51 (20°C)) and like ketones; acetonitrile (relative dielectric constant: 37.5 (20°C)),

- propionitrile (relative dielectric constant: 29.7 (20°C)), butyronitrile (relative dielectric constant: 20.3 (20°C)), isobutyronitrile (relative dielectric constant: 20.4 (20°C)), benzonitrile (relative dielectric constant: 25.2 (25°C)) and like nitriles; formaldehyde, acetaldehyde,
- propionaldehyde and like aldehydes; ethylene glycol monomethyl ether, ethylene glycol monoethyl ether and like alkylene glycol mono alkyl ethers; tetrahydrofuran (relative dielectric constant: 7.6 (25°C)), dioxane (relative dielectric constant: 2.2 (25°C)) and like
- ethers, trifluoroethanol, pentafluoropropanol, 2,2,3,3tetrafluoro propanol and like fluorine alcohols,
 sulfolane (relative dielectric constant: 43.3 (20°C)),
 nitromethane (relative dielectric constant: 35.87 (30°C))
 and the like.
- The relative dielectric constant of water is 78.3 (25°C).

The content of HF is about 0.01-50% by weight, preferably about 1-5% by weight.

The water content is not higher than 70% by 25 weight, preferably not higher than 30% by weight, more

preferably about 0-5% by weight.

The content of the inorganic acid is about 1-99% by weight, preferably about 30-70% by weight.

The content of the organic acid is about 30-5 99.9% by weight, preferably about 70-99.9% by weight.

The content of the organic solvent having a hetero atom is about 30-99.9% by weight, preferably about 70-99.9% by weight.

The content of at least one member selected

from the group consisting of the inorganic acid, organic acid and organic solvent having a hetero atom is about 30-99.9% by weight, preferably about 70-99.9% by weight.

The inorganic acid has a pKa at 25°C of about 2 or lower, preferably about -5 or lower.

The relative dielectric constant of the organic acid and organic solvent having an hetero atom is preferably about 40 or lower, more preferably about 10 or lower.

As the HF is usually used dilute hydrofluoric 20 acid (50 wt. % aqueous solution). However, when the HF does not contain water, 100% HF may be also used.

In case of HCl, HBr and HI, an anhydrous etching solution can be prepared by blowing these gases through the etching solution.

25 Preferable etching solutions of the present

invention and their compositions are shown below.

- •HF: IPA: water = 1-10% by weight: 70-99% by weight: 0-30% by weight
- •HF: acetic acid : water = 0.5-5% by weight : 70-99.5% by
- 5 weight: 0-30% by weight
 - •HF: HCl: water = 0.01-5% by weight: 1-36% by weight: 50-99% by weight
 - ·HF: nitric acid: water = 0.01-5% by weight: 1-70% by weight: 20-99% by weight
- 10 •HF: acetone: water = 1-10% by weight: 70-99% by weight: 0-30% by weight
 - •HF: THF: water = 1-10% by weight: 70-99% by weight: 0-30% by weight
 - •HF: methanol : water = 1-10% by weight : 70-99% by
- 15 weight: 0-30% by weight
 - •HF: ethanol : water = 1-10% by weight : 70-99% by weight : 0-30% by weight

The etching solution of the invention can be suitably used for selectively etching a doped oxide film of an article to be etched comprising an oxide film (BSG, BPSG, etc.) doped with B, P and the like and an undoped oxide film such as THOX, TEOS and like.

In the etching method of the present invention, the temperature of the etching solution is about $15-40\,^{\circ}\text{C}$.

25 Examples of the article to be etched include

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single crystalline silicon wafers, gallium-arsenic wafers and like wafers, especially the articles comprising a doped oxide film (BSG, BPSG, etc.) and an undoped oxide film (THOX, TEOS and like USGs).

The BSG etch rate of the etching solution of the invention is usually about 10-2000 nm/min, preferably about 40-500 nm/min.

The present invention can provide an etching solution which can selectively etch films doped with impurities, such as BSG, BPSG and the like, relative to THOX, TEOS and like USG, a method for producing an etched article using the etching solution and an etched article.

BEST MODE FOR CARRYING OUT THE INVENTION

The present invention will be explained in more detail with referring to Examples and Comparative Examples below.

Examples 1-2 and Comparative Examples 1-4 (inorganic acid)

Etching solutions were prepared by mixing HF,

water, an organic solvent having a hetero atom (isopropyl alcohol (IPA), THF, acetone, methanol, ethanol), an organic acid (acetic acid) and inorganic acid (HCl, HNO3) in the ratios shown in Table 1. Test substrates were produced by forming each of a thermal oxide (THOX) film,

USG (TEOS) film, boron silicate glass (BSG) film and

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boron phosphosilicate glass (BPSG) film on a silicon substrate by CVD method using a tetraethoxysilane gas. The etch rate and etch selectivity of the etching solutions on the test substrates were determined.

In addition, the etch rate and selectivity of conventional HF-H $_2$ O and HF-NH $_4$ F-H $_2$ O etching solutions were determined in the above-mentioned manner as Comparative Examples.

The etch rate was determined by measuring the thickness of the films before and after etching with an Auto EL-III ellipsometer manufactured by Rudolf Research.

The etch rates of the etching solutions were calculated as the difference in thickness of films before and after being etched at 25°C divided by etch time.

The results of the etching solutions with each composition are shown in Table 1 to Table 8.

The relative dielectric constant is that of a solvent (an organic solvent having a hetero atom or an organic acid) + water at 25°C, expressed as a calculated value of an average of the relative dielectric constants of the solvent and water having the particular composition.

Average of relative dielectric constants = [78.3 x (percentage by weight of water) + (relative dielectric constant of solvent at 25°C) x (percentage by

weight of solvent)] / [(percentage by weight of water) +
(percentage by weight of solvent)]

HF-H2O-isopropyl alcohol (PA) etchant

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HF-H20-acetic acid etchant

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TEOS	etch	rate (A/mi	n.)		14	18	22	33	45	55	140	23	32	46	28	65	72
THOX	etch	rate (A/mi	n.)		10	12	17	25	32	40	76	18	20	32	39	40	43
Relative	dielectric	constant of solvent		acid) + water (calculat-	6.88	7.06	7.25	7.62	œ	8.38	9.95	08.6	13.5	20.8	28.1	35.4	42.7
Solvent	(acetic	acid) con-	(%)		86	97.5	76	96	95	94	06	93.75	88.75	78.75	68.75	58.75	48.75
Water	concen-	tration (%)	2		1	1.25	1.5	7	2.5	т	Ŋ	S	10	20	30	40	50
HF con-	centra-	tion (%)			1	1.25	1.5	2	2.5	ю	5	1.25	1.25	1.25	1.25	1.25	1.25
Relativ	Φ	dielect	constan	t of solvent	6.15	6.15	6.15	6.15	6.15	6.15	6.15	6.15	6.15	6.15	6.15	6.15	6.15
Solvent					Acetic	acid Acetic	acid Acetic acid										
	•				Ex. 9	Ex. 10	Ex. 11	Ex. 12	Ex. 13	Ex. 14	Ex. 15	Ex. 16	Ex. 17	Ex. 18	Ex. 19	Ex. 20	Ex. 21

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HF-H2O-tetrahydrofurane (THF) etchant

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HF-H20-acetone etchant

HF Water
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HF-H20-methanol etchant

										_
BPSG	/TEO	ഗ	sele	ctiv	ity	ı		10	26	9.5
BSG/	TEOS	sele	ctiv	ity				6.3	19	17
BPSG	/THO	×	sele	ctiv	ity			150	77	19
BSG/	THOX	sele	ctiv	ity				88	57	33
BSGB	etch	rate	(A/m	in.)				73	230	410
BSG	etch	rate	(A/m	in.)				44	170	730
TEOS	etch	rate	(A/m	in.)				7	o	43
THOX	etch	rate	(A/m	in.)				0.5	m	22
Relative	dielectric	constant of	solvent	(metha-	nol)+water	(calculated	value)	34.0	35.0	39.9
Solvent	(metha-	nol)	concen-	tration	(%)			94	06	80
Water	-uoɔ	cen-	tra-	tion	(%)			3	ഹ	10
ΗF	con-	cen-	tra-	tion	010			3	വ	10
Relative	dielectric	constant	of solvent					32.6	32.6	32.6
Solvent								Ex. 30 Methanol	Ex. 31 Methanol	Ex. 32 Methanol
								Ex. 30	Ex. 31	Ex. 32

HF-H2O-ethanol etchant

tric con- ant cen- vent tra- tion (%)	BSG etch rate (A/m in.)	5 90 27.4 7 9 250 210 36 30 28 23
HF con- cen- tra- tion (%)	Sol- vent (eth- anol) con- cen- tra- trion (%)	06
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 $\mathrm{HF-NH_4F-H_2O}$ etchant (Comparative Examples)

984B	/TEO	ഗ	sele	ctiv	ity		1	I	ı	ì	ı	1	ı	
BSG/	TEOS	sele	ctiv	ity			0.5	1.3	4.0	9.0	0.4	0.4	0.4	
BPSG	/THO	×	sele	ctiv	ity		ı	ı	-	-	_	ı	ı	
BSG/	THOX	sele	ctiv	ity			9.0	2.2	1.4	6.0	9.0	9.0	2.0	
BPSG	etch	rate	(A/m	in.)			1	1	1	ł	ı	1	t	
BSG	etch	rate	(A/m	in.)			110	620	440	350	270	230	200	
TEOS	etch	rate	(A/m	in.)			230	480	640	700	720	610	450	
THOX	etch	rate	(A/m	in.)			170	280	320	400	420	390	300	
Sol-	vent	(water	con-	cen-	tra-	tion (%)	59.9	96	66	88	78	68	59.3	
NH4 F	con-	cen-	tra-	tion	(%)		39.1	2	5	10	20	30	38.7	
HF	con-	cen-	tra-	tion	(%)		T	2	2	2	2	2	2	
Relative	dielectric	constant	of solvent				(78.3)	(78.3)	(78.3)	(18.3)	(18.3)	(18.3)	(78.3)	
Solvent							(Water)	(Water)	(Water)	(Water)	(Water)	(Water)	(Water)	
							Comp. Ex. 4	Comp. Ex. 5	Comp. Ex. 6	Comp. Ex. 7	Comp. Ex. 8	Comp. Ex. 9	Comp. Ex.	10

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HF-H₂O-acid-added etchant

BPSG	/TEO	w	sele	ctiv	ity	ı	1	1	1	1	1	ı		ł		114		
BSG/	TEOS	sele	ctiv	ity		14	13	13	14	12	16	5.9		ഹ				
BPSG	/THO	×	sele	ctiv	ity	ı	1	1	ı	ι	ŧ	1		ı				
BSG/	THOX	sele	ctiv	ity		26	23	21	24	19	22	8.7		7.1				
BPSG	etch	rate	(A/m	in.)		1	1	1	i	ı	ı	ı						
BSG	etch	rate	(A/m	in.)		440	1200	2500	4300	4500	5300	850						
TEOS	etch	rate	(A/m	in.)		32	68	200	300	380	340	170						
THOX	etch	rate	(A/m	in.)		17	53	120	180	240	240	120						
Acid	concen	tratio	n (%)			35.9	35.8	35.6	35.5	35.3	9.89	83.3						
Water	-uop	cen-	tra-	tion	(h)	64	63.9	63.9	63.8	63.7	30.4	15.7						
HF	con-	cen-	tra-	tion	(%	0.1	0.25	0.5	0.75		1	1				•		
pKa of	acid					8-	80	8	ω Ι	8	-1.8	2.15		(pKa1)	7.20	(pKa2)	12.4	(pKa3)
Added	acid		****			HC1	HC1	HC1	HC1	HC1	HNO3	H3PO4						
						Ex. 34	Ex. 35	Ex. 36	Ex. 37	Ex. 38	F.x. 39	Comp. Ex.	11					

CLAIMS

- 1. An etching solution comprising hydrofluoric acid, wherein a ratio of an etch rate of a boron silicate glass film (BSG) or boron phosphosilicate glass / an etch rate of a thermal oxide film (THOX) at 25°C is 10 or higher.
 - 2. The etching solution according to claim 1, wherein a solvent in the etching solution has a relative dielectric constant of 61 or lower.
- 3. The etching solution according to claim 1, the solution containing at least one member selected from the group consisting of an organic acid and an organic solvent having a hetero atom.
- 4. The etching solution according to claim 1,

 the solution containing (i) water and (ii) at least one
 member selected from the group consisting of an organic
 acid and an organic solvent having a hetero atom, the
 water being contained in a concentration of 70% by weight
 or lower.
- 5. The etching solution according to claim 1, wherein the weight ratio of HF: isopropyl alcohol: water is 0.1-50% by weight: 30-99% by weight: 0-70% by weight.
- 6. The etching solution according to claim 1, wherein the weight ratio of HF: acetic acid: water is

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- 0.1-50% by weight: 30-99.9% by weight: 0-70% by weight.
- 7. The etching solution according to claim 1, wherein the weight ratio of HF: tetrahydrofuran: water is 0.1-50% by weight: 30-99.9% by weight: 0-70% by weight.
- 8. The etching solution according to claim 1, wherein the weight ratio of HF: acetone: water is 0.1-50% by weight: 30-99.9% by weight: 0-70% by weight.
- 9. The etching solution according to claim 1,

 10 wherein the weight ratio of HF: methanol: water is 0.1
 50% by weight: 30-99.9% by weight: 0-70% by weight.
 - 10. The etching solution according to claim 1, wherein the weight ratio of HF: ethanol: water is 0.1-50% by weight: 30-99.9% by weight: 0-70% by weight.
- 15 11. The etching solution according to claim 1, the solution comprising an inorganic acid.
 - 12. The etching solution according to claim 11, wherein the inorganic acid has a pKa value at 25°C of 2 or lower.
- 20 13. The etching solution according to claim 11, wherein the weight ratio of HF: HCl: water is 0.01-50% by weight: 1-36% by weight: 0-99% by weight.
 - 14. The etching solution according to claim 11, wherein the weight ratio of HF: HNO_3 : water is 0.01-50% by weight: 1-70% by weight: 0-99% by weight.

- 15. A method for producing an etched article by etching an article to be etched with the etching solution as defined in any of claims 1-14.
- 16. An etched article which is obtainable by5 the method of claim 15.

JC18 Rec'd PCT/PTO 2 2 MAY 2001 IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

For: ETCHING SOLUTION, ETCHED ARTICLE AND

Patent

In re patent application of: KEZUKA ET AL.

Serial No.: Unassigned

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Filed: May 22, 2001

METHOD FOR ETCHED ARTICLE

Examiner: Unassigned

Art Unit: Unassigned

Atty. Docket

No.:P07223US00/LRP

CHANGE OF CORRESPONDENCE ADDRESS CUSTOMER NUMBER DESIGNATION

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Respectfully submitted,

Date: May 22, 2001

By: Douglas E. Jackson Registration No.: 28,518

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DEC	CLARATION FOR USA PA (including Design and Natio	ATENT APPLICATION nal Stage PCT) Attorney's Do	cket ID: P07223US00
As a below named inventor, I hereby de My residence, post office address and citiz is listed below) or an original, first and join	clare that:	ome. I believe I am the original first and s	ole inventor (if only one name
on the invention entitled <u>ETCHIN</u>	G SOLUTION, ETCHED ARTIC	CLE AND METHOD FOR ETCHE	ARTICLE
		, the	
x was filed on November	22, 1999 [] and was am	nended on(or)	
[] as U.S. Application No	1.4444	(or)	
[X] as International PCT Appl	ication No. PCT/JP99/06502		
I hereby state that I have reviewed and under to above. I acknowledge the duty to disc	erstand the contents of the above-identified s	specification, including the claims, as amend	ed by any amendment referred eral Regulations, § 1.56.
I hereby claim foreign priority benefits un certificate, or §365 (a) of any PCT Internati also identified below, where priority is not c date before that of the application on whice	onal application which designated at least o laimed, any foreign application for patent or	ne country other than the United States of A	merica. Tisted below and have
Prior Foreign Application(s) (AD	DITIONAL APPLICATIONS IDENTIFIED	O ON ATTACHED SHEET):	
Number 1998–332782	Country Japan	Day/Month/Year Filed 24/11/1998	Priority Not Claimed
I hereby claim the benefit under Title 35, Un the U.S., listed below; and insofar as the stapplication in the manner provided by the ft to patentability as defined in Title 37, Code or PCT international filing date of this app	abject matter of each of the claims of this a first paragraph of Title 35, United States Coo of Federal Regulations 8 1.56 which because	upplication is not disclosed in the prior Unit ie, § 112, I acknowledge the duty to disclose the available between the filing date of the no	ed States or PCT International information which is material or application and the national
Application Serial No.	Day/Month/Year Filed	Status pate	nted, pending, abandoned
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SIGN AND DATE HERE: Inventor's Signature:	Takehiko Kauka	Date: Ma	y 9, 2001
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SIGN AND DATE HERE: Inventor's Signature:	makoto Suyama	Date: N	May 9, 2001
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	ame as the above		/
SIGN AND DATE HERE: Inventor's Signature:	M. Taux	Dato: N	May 9, 2001
Full Name of Fourth		Citizenship	
Joint Inventor, if any			

Residence - City, State/Country
(if different from P.O. address)

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DATE HERE: Inventor's Signature:

______ SEE ATTACHED SHEET FOR SIMILAR INFORMATION AND SIGNATURE FOR ADDITIONAL JOINT INVENTORS.

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